

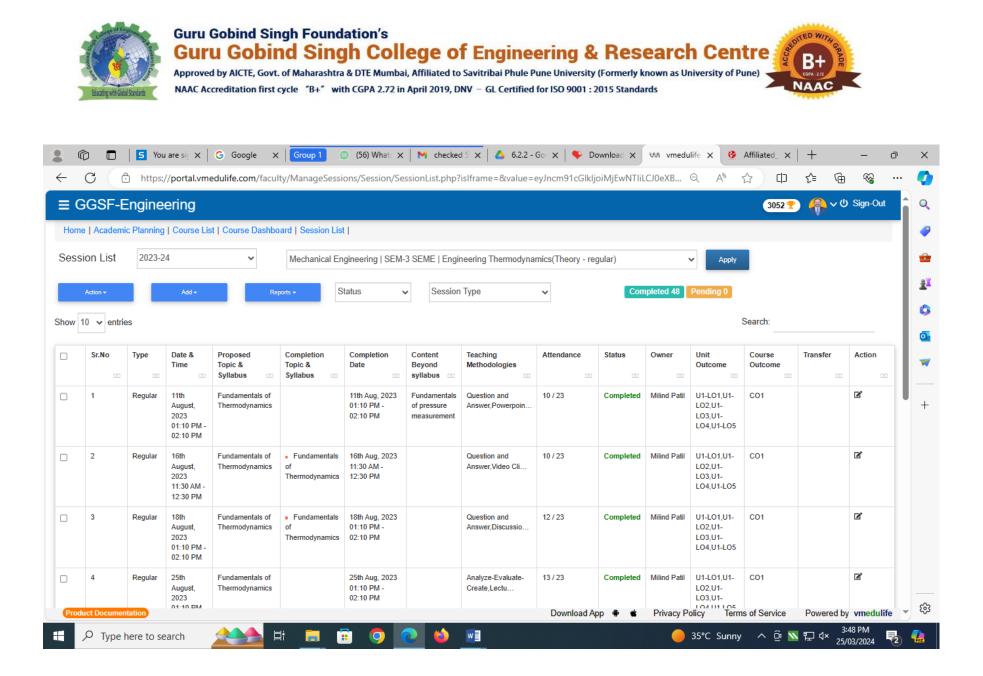
## 7.2.1 - Describe two best practices successfully implemented by the Institution as per NAAC format provided in the Manual.

Title - Digitization of academic processes with focus on implementation of OBE (Few Sample Screen shots)

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Home   Academic	Planning   Course List					
Course List		A Manage Courses				
2023-24	~					
Mechanical Engin	eering				View Workload I	Distribution
Semester / Group	Course					Action
SEM-3 SEME	Engineering Thermodynamics ( Theory )	🔁 48 Sessions	🗈 6 Units	G→ 6 Course Outcome	स्र Attendance Report	۲
	Engineering Thermodynamics ( Practical ) Batch: SEME-A1	ia 8 Sessions	🗈 8 Units	G→ 5 Course Outcome	R Attendance Report	۲
	Engineering Thermodynamics ( Practical ) Batch: SEME-A2	2 8 Sessions	🗈 8 Units	G 5 Course Outcome	R Attendance Report	۲
SEM-7 BEME	Heating Ventilation Air-Conditioning and Refrigeration ( Theory )	27 Sessions	🗈 6 Units	G⇒ 6 Course Outcome	R Attendance Report	۲
	Heating Ventilation Air-Conditioning and Refrigeration ( Practical ) Batch: BEME-A1	D 9 Sessions	🗈 8 Units	G⇒ 5 Course Outcome	R Attendance Report	۲
	Heating Ventilation Air-Conditioning and Refrigeration ( Practical ) Batch: BEME-A2	2 8 Sessions	🗈 8 Units	G⇒ 5 Course Outcome	R Attendance Report	۲
	Heating Ventilation Air-Conditioning and Refrigeration ( Practical ) Batch: BEME-A3	a Sessions	🗈 8 Units	G 5 Course Outcome	स्र Attendance Report	۲
	Heating Ventilation Air-Conditioning and Refrigeration ( Practical ) Batch: BEME-A4	₽ 8 Sessions	🗈 8 Units	G→ 5 Course Outcome	R Attendance Report	۲
PROJECT	Project-I Group 03: MSP ( Practical ) Batch: Group-03: MSP	₽ 37 Sessions	🗈 3 Units	G→ 3 Course Outcome	A Attendance Report	۲
SEM-8: BEME	Energy Engineering ( Theory )	27 Sessions	E 6 Units	G→ 6 Course Outcome	R Attendance Report	۲









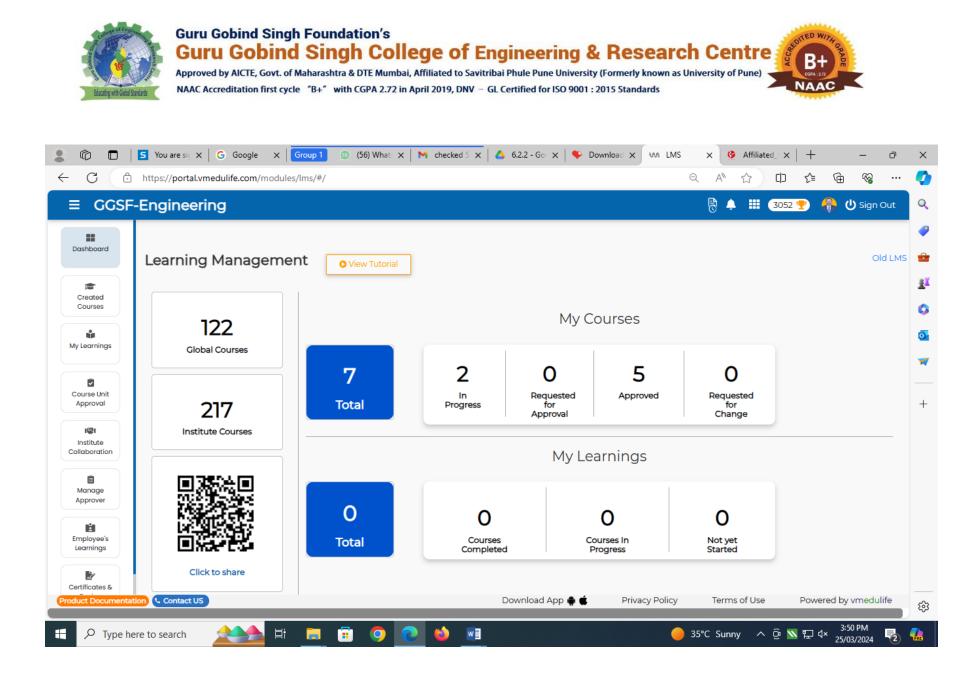




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👧 Guru Gobind Singh College of Engineering & Research Centre, Nashik	
Stream : Mechanical Engineering	
Title: MCQ Test Unit 1 Subject: Energy Engineering - Theory   Faculty: Milind Patil Academic Year: 2023-24   Year: FINAL YEAR - SEM-8: BEME   Negative Marking: Not Applicable farks: 20   Date: 2024-02-02   Duration: 60 minutes	+
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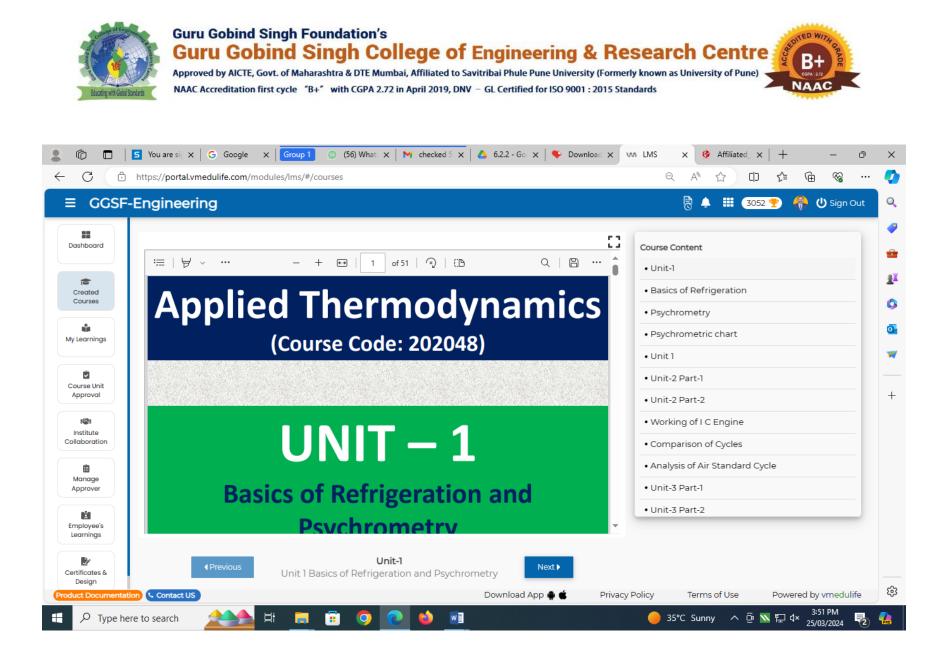




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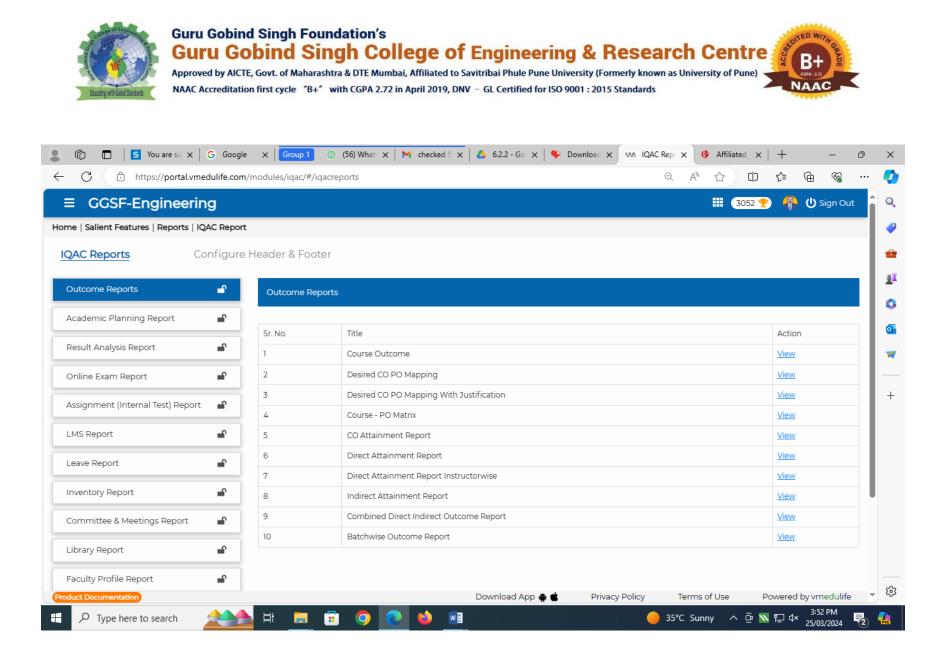






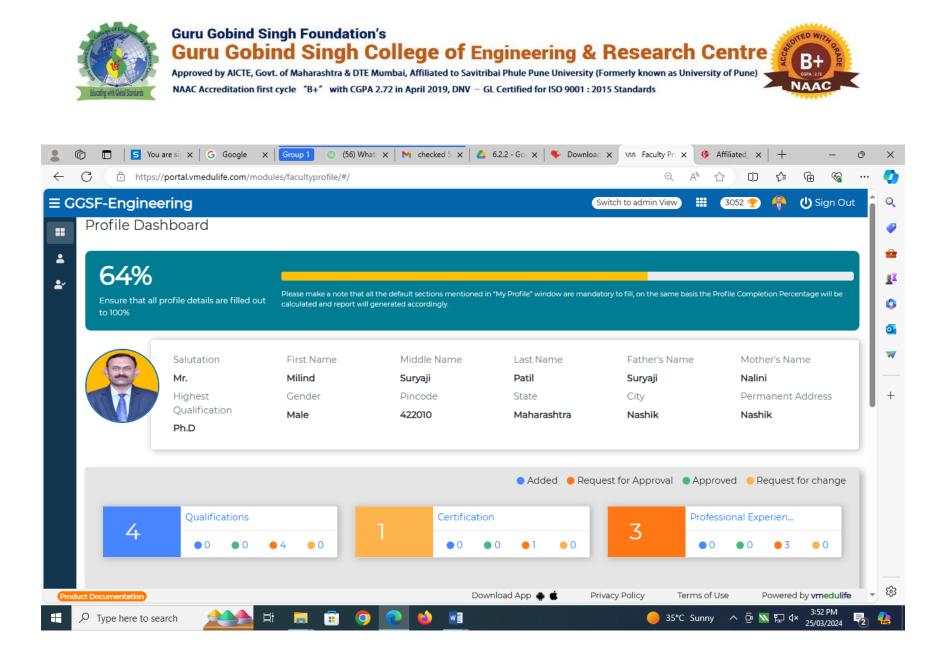






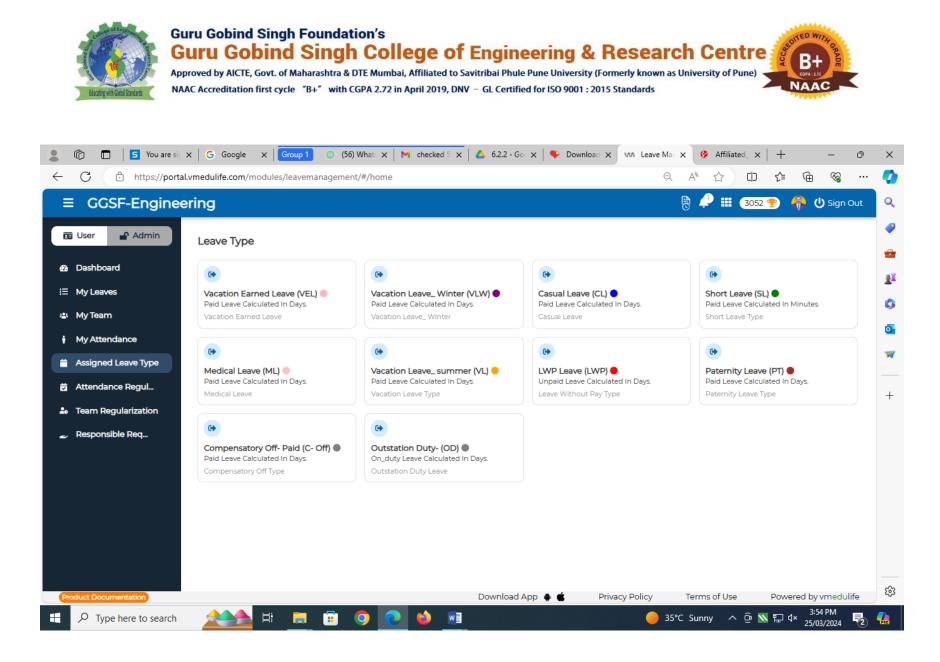
















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Grade Students for Assignment Basic Learning		×
E View Student List Roll No: 12 - Neha Kurnawat		ts
Grade Student	Student's Assignment	
A can of soft drink at room temperature is put into the refrigerator so that it will cool. Would you model the can of soft drink as a closed system or as an open system? Explain.	Assignment No.1	Q
Enter Marks 3	RI) A can of soll drink at room temp. Is pat referigerator so that it will cool woul Can of soft drink as close system or op Explains	10 900 million -
What is a quasi-equilibrium process?     Mark(s) 3.00 Type       What is its importance in engineering?     DESCRIPTIVE	-> Here, I'll model a can of soft drink int	to a clase
Consider a system whose temperature is 18°C. Express this temperature in R, K, and °F.	System. - In close system energy can exchange its symmounding Cheat of can exchange but their is no mass transfer equant Soft drink remain same) making it a	tity of
The temperature of a system rises by Type Type	Soft drink remain same making it	





